

Unusual reaction of aziridine dimer with acetylene dicarboxylates

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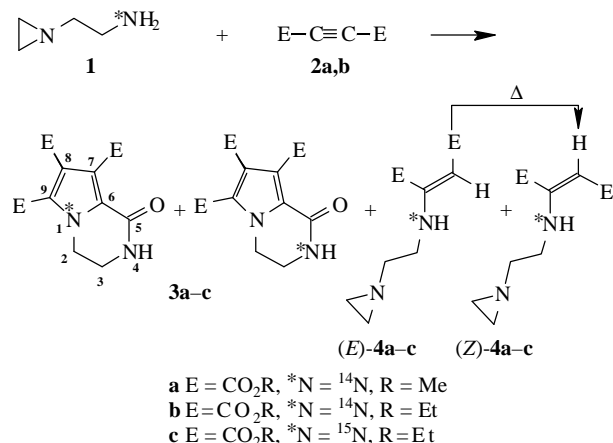
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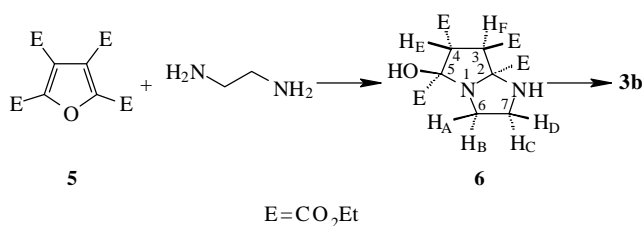
The 1-(2-aminoethyl)aziridine (aziridine dimer) **1** reacts with acetylene dicarboxylates to give 1,4-diaza-5-oxobicyclo[4.3.0]nonanes (7-azaindolizines) **3** together with the usual enamine adducts **4**; the product structures are confirmed spectroscopically and, in the **3** also by X-ray diffraction and by an independent synthesis from furan **5**.

Treatment of the aziridine dimer **1**[†] and the terminally labelled dimer **1**-¹⁵NH₂[†] with acetylene dicarboxylates **2a,b** afforded the 7,8,9-tris(alkoxycarbonyl)-1,4-diaza-5-oxobicyclo[4.3.0]nonanes **3a-c**, along with the usual adducts (*E*)- and (*Z*)-**4a-c** (Scheme 1).

The structures of **3,4** were confirmed spectroscopically[‡] (Figure 1). The structure of **3a** was also confirmed by X-ray diffraction[§] (Figure 2) and also by an independent synthesis of **3b** from furan **5**[‡] (Scheme 2). The formation of **3a,b** which is also accompanied by ethylene elimination and ¹⁵N label randomization can be explained by the transformations



Scheme 1



Scheme 2

[†] NMR spectra were recorded at 400 MHz (¹H) and 100.61 MHz (¹³C). The aziridine dimer **1**-¹⁵N was prepared by the reduction of aziridinoacetamide-¹⁵NH₂ (LiAlH₄ in boiling THF, 12h), yield 12.3%, bp 127 °C; ¹H NMR (CDCl₃): δ 1.13 (m) and 1.72 (m) (2×2H, CH₂ ring), 1.43 (br. s, 2H, HN), 2.26 (dt, 2H, α-CH₂, ³J_{HH} 5.8, ³J_{HN} 2.1 Hz) and 2.85 (dt, 2H, β-CH₂, ²J_{HN} 0.9 Hz); ¹³C NMR (CDCl₃): δ 26.6 (dd, CH₂ ring, ¹J 163.7 and 173.4 Hz), 41.79 (ddt, β-CH₂, ¹J 133.2, ²J 2.8, ¹J_{CN} 4.2 Hz) and 64.36 (dt, α-CH₂, ¹J 133.2, ²J_{CN} 1.4 Hz). The starting material aziridinoacetamide-¹⁵NH₂ was prepared from ethyl aziridinoacetate² by amidation with ¹⁵NH₃ (90% of ¹⁵N) in MeOH–MeONa, 7 days at 20 °C; yield 89.4%, mp 88 °C; ¹H NMR (CDCl₃): δ 1.27 (m) and 1.84 (m) (2×2H, CH₂ ring), 2.90 (s, CH₂CO) and 6.72 (2H, H_AH_B, ¹⁵N, ABX spectrum, Δν = 200, ²J_{AB} 4.3, ¹J_{BX} 89.5, ¹J_{BX} 90.3 Hz); ¹³C {¹H} NMR (CDCl₃): δ 27.4 (s, CH₂ ring), 63.4 (d, CH₂CO, ²J_{CN} 5.6 Hz) and 173.0 (d, CO, ¹J_{CN} 15.3 Hz).

indicated in Scheme 3. The key stages of the proposed mechanism are confirmed by the known easy ethylene

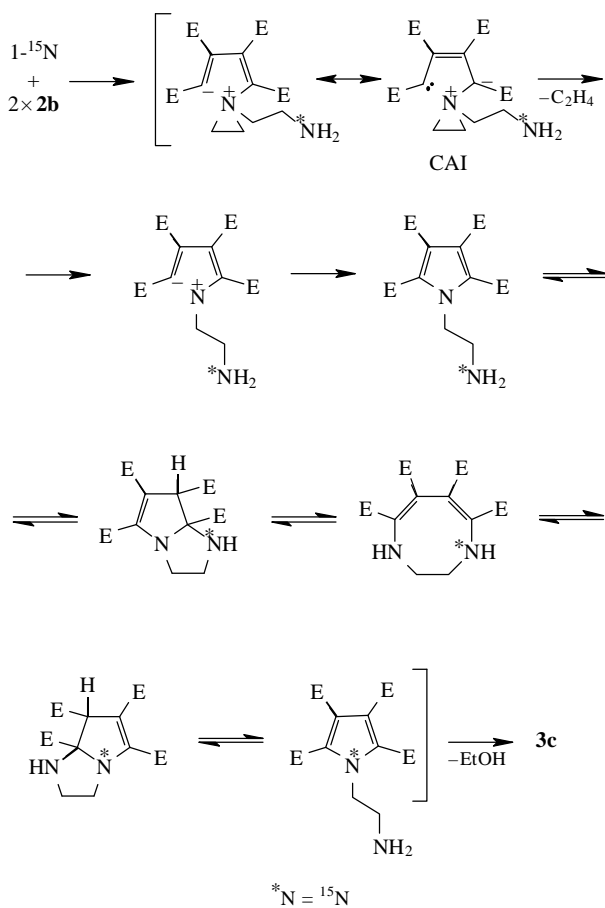
[‡] Spectroscopic data for **3a**: obtained from **1** and **2a** in Et₂O at 20 °C, 1h; yield 19.1%, mp 188–189 °C. Found: C, 50.41; H, 4.48; N, 9.12. Calc. for C₁₃H₁₄N₂O₇: C, 50.32; H, 4.55; N, 9.03%. ¹H NMR ([²H₈]toluene): δ 2.73 (m, 2H, 3-CH₂), 3.62 (m, 2H, 2-CH₂), 3.54 (s), 3.62 (s) and 3.71 (s) (3×3H, MeO), and 7.30 (br. s, 1H, HN); ¹³C NMR ([²H₈]DMSO): δ 38.52 (t, 3-CH₂N, ¹J 141.0 Hz), 43.65 (2-CH₂, ¹J 146.4 Hz), 51.56 (q), 52.05 (q) and 52.36 (q) (3MeO, ¹J 147.7 Hz), 118.3 (s), 119.2 (s) (7- and 8-C), 122.64 (s, 9-C) and 125.58 (dt, 6-C, ³J_{CNH} 2.7, ³J_{CNH} 5.6 Hz), 156.81 (q, 5-C, ²J = ³J = 4.2 Hz), 159.32 (q), 162.88 (q) and 163.29 (q) (CO₂Me, ³J 4.2 Hz); MS (EI, 70 eV, 240 °C): *m/z* (%): 310 (34.3) [M⁺], 280 (15), 279 (100), 59 (13.2).

For **3b**: obtained from **1** and **2b** in Et₂O at 20 °C, 18 h, yield 31%, or quantitatively from **6** in [²H₈]toluene in a sealed tube at 140 °C, 1h; mp 140–141 °C. Found: C, 54.62; H, 5.68; N, 8.10. Calc. for C₁₆H₂₀O₇N₂: C, 54.54; H, 5.72; N, 7.95%. ¹H NMR ([²H₈]toluene): δ 1.08 (t), 1.13 (t) and 1.21 (t) (3×3H, Me, ³J 7 Hz), 2.47 (br. m, 2H, 3-CH₂), 3.53 (br. m, 2H, 2-CH₂), 4.07 (q), 4.19 (q) and 4.33 (q) (3×2H, 3CH₂O) and 7.41 (br. s, 1H, HN); ¹³C NMR (CD₃OD): δ 14.32 (q, 2Me, ¹J 126.2 Hz), 14.38 (Me, ¹J 127.4 Hz), 40.43 (tt, 3-CH₂, ¹J 142.9, ²J 2.8 Hz), 45.14 (tt, 2-CH₂, ¹J 147.0, ²J 2.8 Hz), 62.64 (tq, 2CH₂O, ¹J 148.4, ²J 4.2 Hz), 62.9 (tq, CH₂O, ¹J 148.2, ²J 4.2 Hz), 120.79 (s), 122.65 (s) and 124.6 (s) (9-, 7-, 8-C), 127.02 (m, 6-C, ³J 2.8 Hz), 159.96 (t, 5-C, ²J 4.2 Hz), 160.73 (m), 165.11 (m) and 165.15 (m) (CO₂Et, ³J 4.2 Hz).

For **3c**: yield 12.1%, mp 143–144 °C (PrⁱOH); ¹H NMR ([²H₈]toluene): 1.09 (t), 1.14 (t) and 1.23 (t) (3×3H, 3Me, ³J 7.0 Hz), 2.45 (br. m, 2H, 3-CH₂, ²J_{HN} 1.8 Hz), 3.5 (br. m, 2H, 2-CH₂, ³J_{HN} 2.1 Hz), 4.08 (q), 4.20 (q) and 4.33 (q) (3×2H, 3CH₂O) and 6.65 (d, 1H, HN, ¹J_{HN} 91.6 Hz); ¹³C NMR (CD₃OD): 14.31 (q, 2Me, ¹J 126.2 Hz), 14.42 (Me, ¹J 127.4 Hz), 40.43 (ttd, 3-CH₂, ¹J 142.9, ²J_{CH} 2.8, ¹J_{CN} 8.4 Hz), 45.14 (ttd, 2-CH₂N, ¹J 147.0, ²J_{CH} 2.8, ¹J_{CN} 7.7 Hz), 62.61 (tq), 62.62 (tq) and 62.9 (tq) (3CH₂O, ¹J 148.2 Hz), 120.8 (dd, 7-C, ²J = ³J_{CN} = 2.9 Hz), 122.63 (d, 8-C, ¹J_{CN} 4.5 Hz), 124.58 (d, 9-C, ¹J_{CN} 14.6 Hz), 126.96 (d, 6-C, ¹J_{CN} 4.5 Hz), 159.93 (d, 5-C, ³J_{CN} 16.7 Hz), 160.73 (m), 165.08 (m) and 165.14 (m) (CO₂Et, ³J 4.2 Hz).

For **4a**: (*Z*: *E*, 3:1), yield 36.3%. For (*Z*)-**4a**: ¹H NMR (CDCl₃): δ 1.12 (m) and 1.75 (m) (4H, 2CH₂ ring), 2.34 (t, 2H, α-CH₂, ³J 6.0 Hz), 3.53 (td, β-CH₂, ³J_{HCNH} 6.0 Hz), 3.65 (s) and 3.81 (s) (2×3H, 2MeO) (br. s, 1H, HN). For (*E*)-**4a**: ¹H NMR (CDCl₃): δ 1.15 (m) and 1.74 (m) (4H, 2CH₂ ring), 2.41 (t, 2H, α-CH₂, ³J 6.0 Hz), 3.13 (td, 2H, β-CH₂, ³J_{HCNH} 6.0 Hz), 3.62 (s) and 3.86 (s) (2×3H, 2MeO), 4.64 (s, 1H, HC=) and 5.3 (br. s, 1H, HN).

For **4b, 4c**: (*Z*: *E*, 1.2:1), yields 27%. (*Z*)-**4b**: ¹H NMR (CDCl₃): δ 1.12 (m) and 1.74 (m) (4H, 2CH₂ ring), 1.24 (t) and 1.34 (t) (2×3H, 2Me, ³J 7.0 Hz), 2.34 (t, 2H, α-CH₂, ³J 6.1 Hz), 3.53 (td, β-CH₂, ³J_{HCNH} 6.1 Hz), 4.07 (q) and 4.25 (q) (2×2H, 2CH₂O), 5.05 (s, 1H, HC=), 8.30 (br. s, 1H, HN). (*Z*)-**4c**: ¹H NMR (CDCl₃): δ 1.12 (m) and 1.74 (m) (4H, 2CH₂ ring), 1.24 (t) and 1.33 (t) (2×3H, 2Me, ³J 7.0 Hz), 2.34 (dt, 2H, α-CH₂, ³J_{HH} 6.1, ³J_{HN} 2.8 Hz), 3.53 (tdd, 2H, β-CH₂, ²J_{HCN} ca. 1, ³J_{HCNH} 6.1 Hz), 4.08 (q) and 4.24 (q) (2×2H, 2CH₂O), 5.04 (d, 1H, HC=, ³J_{HN} 4.0 Hz) and 8.26 (dt, 1H, HN, ¹J_{HN} 90.4 Hz). (*E*)-**4b**: ¹H NMR (CDCl₃): δ 1.16 (m) and 1.74 (m) (4H, 2CH₂ ring), 1.23 (t) and 1.33 (t) (2×3H, 2Me, ³J 7.0 Hz), 2.41 (t, 2H, α-CH₂, ³J 6.0 Hz), 3.13 (td, β-CH₂, ³J_{HCNH} 6.0 Hz), 4.10 (q) and 4.30 (q) (2×2H, 2CH₂O), 4.64 (s, 1H, HC=) and 5.09 (br. s, 1H, HN). (*E*)-**4c**: ¹H NMR (CDCl₃): δ 1.15 (m) and 1.74 (m) (4H, 2CH₂ ring), 1.23 (t) and 1.35 (t) (2×3H, 2Me, ³J 7.0 Hz),



Scheme 3

elimination from aziridinium ylides⁴ [the first intermediate zwitterion in Scheme 3 can be represented as a mesomeric form of carbenaziridinium ylide (CAI)] and by the isolation of

2.41 (br. m, 2H, α -CH₂), 3.12 (tdd, β -CH₂, $^2J_{\text{HCNH}}$ 6.1 Hz), 4.10 (q) and 4.29 (q) (2 \times 2H, 2CH₂O), 4.62 (d, 1H, HC=, $^3J_{\text{HN}}$ 1.5 Hz), 5.37 (dt, 1H, HN, $^1J_{\text{HN}}$ 93 Hz).

Furan **5**, under the action of ethylene-1,2-diamine in toluene (14 days, 20°C) gave intermediate **6**. The latter was isolated by gradient chromatography (silica gel, heptane–ethyl acetate, 0→60%), yield 24%, colourless oil. Found: C, 52.10; H, 6.69; N, 6.82. Calc. for C₁₈H₂₈N₂O₉: C, 51.92; H, 6.78; N, 6.76%. ¹H NMR (CDCl₃): δ 1.11 (t), 1.13 (t) and 1.20 (t) (3 \times 3H, 3Me, 3J 7.0 Hz), 2.29 (dt, 1H, H_A, $^2J_{\text{AB}}$ -10.4, $^3J_{\text{AC}}$ 10.4, $^3J_{\text{AD}}$ 5.8 Hz), 2.83 (dt, 1H, H_B, $^3J_{\text{BC}}$ 5.8, $^3J_{\text{BD}}$ 10.4 Hz) and 3.02 (ddd, 1H, H_C, $^2J_{\text{CD}}$ -11.6 Hz), 3.08 (ddd, 1H, H_D), 3.43 (d, 1H, H_F, 3J 12.5 Hz), 4.32 (1H, H_E), 4.03 (m), 4.11 (m) and 4.13 (m) (3 \times 2H, 3CH₂); ¹³C NMR (CDCl₃): δ 13.62 (q), 13.71 (q) and 13.81 (q) (3Me, 1J 127.9 Hz), 45.36 (dd, 7-C, 1J 138.1 and 143.9 Hz), 48.22 (dd, 6-C, 1J 138.1 and 143.9 Hz), 52.14 (dd, 3-C, 1J 133.7, 2J 4.4 Hz), 54.48 (dd, 4-C, 1J 138.1, 2J 5.8 Hz), 89.81 (m) and 89.87 (m) (2-C and 4-C, 2J , 3J 4.4 Hz), 168.14 (m) and 169.54 (m) (CO, 2J 7.3, 3J 4.4 Hz), 171.36 (m) and 172.1 (m) (C=O, 3J 4.4 Hz).

[§] Crystal data for **3a**: C₁₃H₁₄N₂O₇ (from PrⁱOH): monoclinic, space group *P*2₁/*n*, at 293 K: *a* = 10.222(12), *b* = 7.903(8), *c* = 18.07(2) Å, β = 102.88°, *V* = 1423(3) Å³, *Z* = 4, *d*_{calc} = 1.448 g cm⁻³, μ = 1.19 cm⁻¹, *F*(000) = 648. Intensities of 2760 reflections were measured with a Siemens P3/PC diffractometer at 293 K (MoK α radiation, graphite monochromator, $\theta/2\theta$ -scan method, 2θ < 50°), and 1480 independent observed ones with *I* > 2 σ (*I*) were used in the further calculations. The structure was solved by direct method and refined by the least-squares against *F* in an anisotropic-isotropic (H-atoms) approximation to *R* = 0.065, \overline{wR} = 0.062, GOF = 1.69. All calculations were performed using the program package SHELXTL PLUS (ver. 5). Full lists of bond lengths, bond angles, atomic coordinates and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre (CCDC). See Notice to Authors, *Mendelev Commun.*, 1997, issue 1. Any request to the CCDC for this material should quote the full literature citation and the reference number 1135/13.

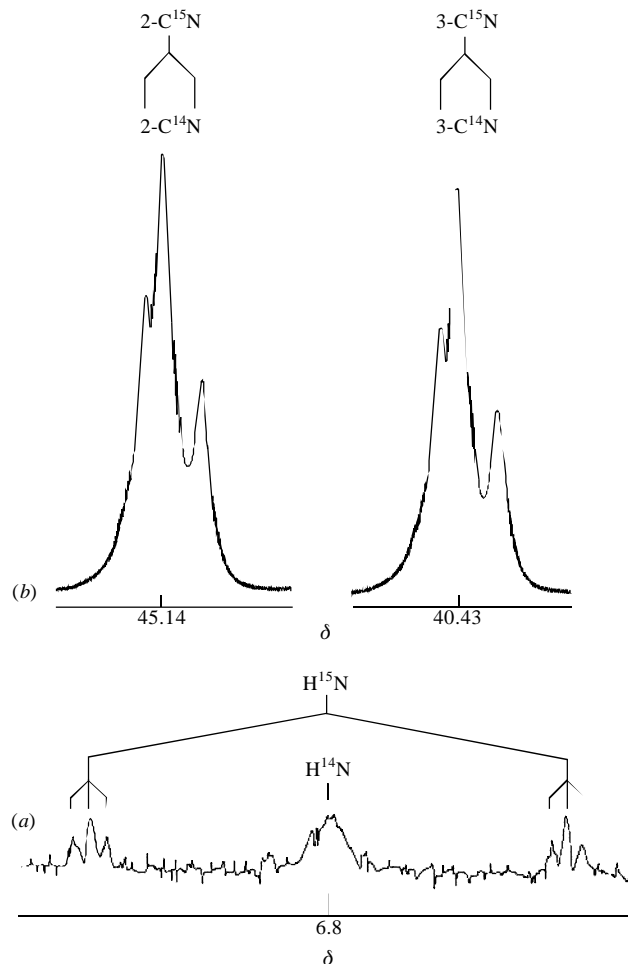


Figure 1 (a) ¹H NMR spectrum of the 4-NH proton of **3c** and (b) ¹³C {¹H} NMR spectrum of the 2-C and 3-C carbon atoms of **3c**.

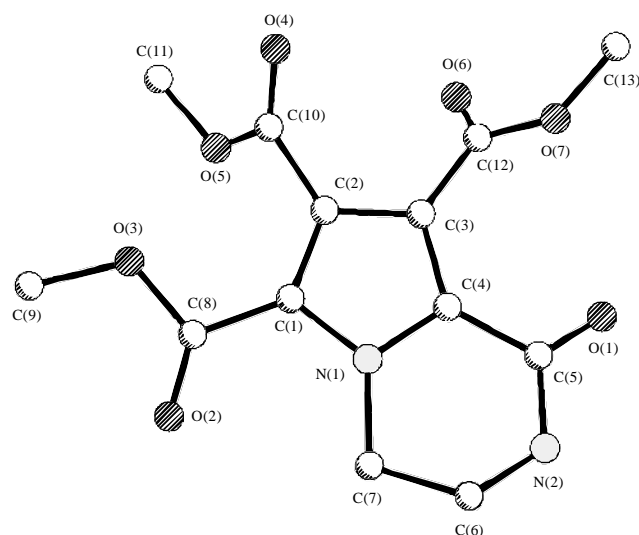


Figure 2 Crystal structure of **3a**. Selected bond lengths (Å) and angles (°): N1–C1 1.383(6), C1–C2 1.385(6), C2–C3 1.401(6), C3–C4 1.402(6), C4–N1 1.373(6), C4–C5 1.477(6), C5–N2 1.347(6), N2–C6 1.459(7), C6–C7 1.509(7), C7–N1 1.476(5); C1N1C4 109.7(3), N1C1C2 106.7(4), C1C2C3 109.2(4), C2C3C4 106.5(4), C3C4N1 107.9(4), C4N1C7 122.5(4), N1C4C5 120.2(4), C4C5N2 114.6(4), C5N2C6 122.6(4), N2C6C7 111.6(4), C6C7N1 107.7(4), C1N1C7 127.8(4), C3C4C5 131.3(4).

the intermediate **6** before the lactamization in **3b** (Scheme 2).

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References

- 1 R. G. Kostyanovsky, V. P. Leshchinskaya, R. K. Alekperov, G. K. Kadorkina, L. L. Shustova, Yu. I. El'natanov, G. L. Gromova, A. E. Aliev and I. I. Chervin, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1988, 2566 (*Bull. Acad. Sci. USSR, Div. Chem. Sci.*, 1988, **37**, 2315) and references cited therein.
- 2 H. Bestian, *Ann.*, 1950, **566**, 210.
- 3 I. J. Cantlon, W. Cocker and T. B. H. McMurry, *Tetrahedron*, 1961, **15**, 46.
- 4 Y. Hata and M. Watanabe, *Tetrahedron Lett.*, 1972, 3827, 4659.

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